

Из таблицы видно, что не по всем загрязнениям действующие очистные сооружения обеспечивают требуемую степень очистки. Особенно неудовлетворительна работа очистных сооружений по хлоридам и содержанию красителей. Используемые в настоящее время на очистных сооружениях способы обработки сточных вод основаны на электрокоагуляции и применении химических реагентов (коагулянтов и флокулянтов). При этом эффективность применения полиакриламида, используемого в качестве флокулянта, является крайне низкой. При существующем режиме очистки сточных вод во флотаторах отделение взвешенных веществ малоэффективно, а их осаждению технологических емкостях препятствуют высокие скорости движения сточной вод.

Проведенные исследования эффективности работы очистных сооружений сточных вод текстильного предприятия показали, что они не позволяют обеспечить требуемую степень очистки. Данная проблема должна решаться путем совершенствования существующих технологий очистки сточных вод с применением более эффективных реагентов, фильтрующих материалов для доочистки и оптимизации режима работы очистных сооружений [2].

#### СПИСОК ЦИТИРОВАННЫХ ИСТОЧНИКОВ

Лесович, Е.В. Эффективность очистки сточных вод текстильного производства / Е.В. Лесович // Стратегические проблемы охраны и использования водных ресурсов: материалы IV Международного водного форума, Минск, 12–13 окт. 2010 г. / Минсктиппроект. – Минск, 2011. – С. 297–299.

Лесович, Е.В. Анализ методов очистки сточных вод предприятий текстильной промышленности / Е.В. Лесович // Сб. науч. Тр. / Полесский аграрно-экологический институт НАН Беларуси – Брест, 2010 – Вып. 3: природная среда Полесья: особенности и перспективы развития. – С. 98–100.

УДК 628.33

MAŽEIKIENĖ A. <sup>1</sup>, ŠVEDIENĖ S. <sup>2</sup>

<sup>1</sup> Vilnius Gediminas Technical University, Department of Water Management, Saulėtekio ave. 11, LT-10223 Vilnius, Lithuania

<sup>2</sup> Vilniaus kolegija/University of Applied Sciences, Faculty of Agrotechnologies, Buivydiskiu str. 1, Buivydiskes, LT-14160 Vilnius area, Lithuania

#### EXPERIMENTAL RESEARCH ON SORPTION OF PETROLEUM PRODUCTS FROM STORM WATER BY FILTRATION

**Abstract.** This article describes the analyses, which were accomplished with the help of the device of experimental filtration, when pollutants from wastewater were being removed by filtration wastewater through sorbents FIBROIL<sup>R</sup> filler in filtering rate 30 m/h. The results showed how efficiently and how much time, under described conditions, it can be cleaned by filtrating wastewater which is variously contaminated (based on the concentrations of petroleum products and suspended solids), until the concentration of petroleum products in filtrate will be  $\leq 5$  mg/L.

Lately in Lithuania and other countries, one of the most polluting substances in the depths of the earth, are petroleum products [1, 2, 3,]. This runoff must be cleaned. Filters reduce the separation time of petroleum products, which may lead to reduced capacity and runoff contamination by petroleum products length of stay in it. For this purpose, the petro-

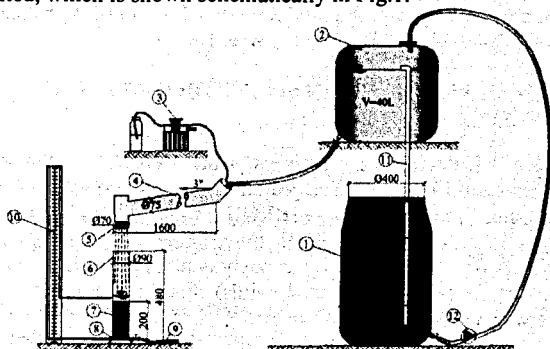
leum trap design included one more stage – i.e. final disposal of petroleum products with sorption filter, which can be used with different sorbents. Application of sorption processes in removing various pollutants from storm water is now the subject of environmental science [2, 4-6]. Also it is important to examine, how much contaminated (based on turbidity and SS) storm water is appropriate to clean by sorbent filtration. The scientific literature contains a lot of information about the application of various sorbents to remove petroleum from the waters surface or aqueous solutions in a static way (collecting petroleum spills, when the concentration of petroleum products is high enough) [2, 7-9]. Scientific experiments are also underway, when the sorbents are used as filter filler, but the filtration rate is low: 0.3-0.4 m/h, [10]; 0.5-1.5 m/h [12].

In reality, the flow of storm water is uneven, constantly changing the concentration of runoff pollutants. To ensure efficient use of absorptive materials, it is necessary to evaluate storm water contamination by suspended solids or turbidity [11, 12] by adjusting filtered runoff content properly, otherwise sorbent acts as a mechanical filter [1, 4].

The purpose of the experiments that have been described in this article – to find out how effective absorption fibrous material FIBROIL<sup>®</sup> removes petroleum products from various pollution levels storm water (according to turbidity, SS and concentration of petroleum products), when runoff is filtered through FIBROIL filler in selected rates.

The object of the researches that have been described in this article – absorption fibrous material FIBROIL<sup>®</sup> and its application possibilities to petroleum products removal from aqueous solutions (compounds) by filtering. It is stated that the FIBROIL<sup>®</sup> is a material which does not get wet and it doesn't absorb water that is why it can be used for filter fillers, when water and petroleum products' compounds are filtered. The scientific literature does not indicate what would be the settings of filtering through fibroil fillers. At the same time FIBROIL<sup>®</sup> is quite expensive material (20 Lt/kg) that is why it is important to explore how storm water characteristics affects the efficiency of petroleum products absorption.

This article describes the smooth way of insertion of petroleum products in filtering runoff or water, when in the filtering device with help of pumps in constant rates (yields) runoff (or water) and petroleum products are supplied. Thus it was possible to prepare compounds with high concentrations of petroleum products (~50, ~100, ~150 mg/L). To achieve above-described objectives, in Vilnius Gediminas Technical University laboratory filtration device was installed, which is shown schematically in Fig.1.



*Fig.1 – Filtering device: 1 - 100 liters receptacle for storm water pouring; 2 - 50 liters tank; 3 - vessel filled with diesel; 4 - pipe, through which run-off with inserted petroleum products are delivered to the filter; 5 - percolator; 6 - filtering cylinder; 7 - fibroil filler; 8 - chip layer which maintain filler; 9 - flexible hose for taking filtrate's samples; 10 - piezometer; 11 - overflow pipe; 12 - pump*

Storm water was poured into the receptacle (1), pump (12) delivers to the tank (2), in which a constant volume of liquid was kept to ensure steady speed of pipe (4) flow with incline (3°). The rate of filtration was regulated by opening bolt more or less and yield of the filtrate was measured every 10 minutes. Petroleum products (diesel) were inserted from the vessel (3) to the flowing liquid by peristaltic pump, at such speed that initial  $C_0$  petroleum products' concentration would form at the end of the pipe (there have been sampled). Further, runoff or tap water was delivered to the cross-sectional area (0,005 m<sup>2</sup>) of filtering cylinder with an initial concentration of petroleum products and with support of percolator and even load. It was filtered through a 20 cm height and 70 grams (70 kg/m<sup>3</sup> filling weight) FIBROIL<sup>®</sup> layer. For these analysis 20, 25 and 30 m/h filtering rate was chosen. Filtrates (of 1 DV) samples and compounds' A and B samples (before filtering cylinder) were collected in jars (0.5 L) every 10 minutes; these samples' contamination was measured by concentrations and turbidity of petroleum products and suspended solids. In filtering cylinder comprising pressure losses was measured with assistance of piezometer (10). Each separate filtering experiment was done with new fibroil filler, weighting and filling density in the cylinder were the same. Measurements of parameter were repeated 3-4 times. Errors of sampling and accuracy of devices measuring constituted errors. Samples were analyzed using standards methods: suspended solids (SS) (LST EN 872:2000), turbidity (LST EN ISO 027:2002) and total petroleum hydrocarbons (TPH) (ISO 9377-2:2000).

During the hydraulic test of experimental stand, pressure losses in fibroil filler, filtering

tap water in filtration rate of 20, 25, 30 m/h was measured. Pressure losses' dependence on filtration speed after 10 minutes (filtering tap water with a turbidity of 1 DV) is shown in Fig. 2.

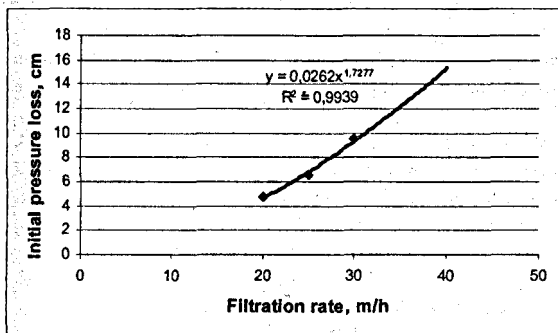


Fig.2 - Dependence on filtering rate in pressure losses filler

As it can be seen from the figure, the pressure losses directly depended on the filtration rate: when the filtration rate was equal to 20 m/h, the pressure losses were up to 5.2 cm; when the filtration rate was equal to 30 m/h, the pressure losses were up to 10.5 cm. In longer filtration of tap water, the pressure losses increased: one hour of filtration on the average increases 0.5 cm.

The filtration of tap water and diesel compound (SS~0.5 mg/L,  $C_0$ ~50 mg/L) in a constant rate of 30 m/h, through 76 g fibroil filler lasted 6 hours. During that time, the concentration of petroleum products in filtrate samples  $C_f$ , varied from 0.5 to 4.5 mg/L and did not exceed the regulated (5mg/L) concentration. In filtration of storm water and diesel compound (SS~10 mg/L), the rate of filtration (the lower, the higher the efficiency) and the ini-

tial petroleum products' concentration (the higher in the limits of analysis, the higher the efficiency) of runoff influenced the efficiency of absorption of fibroil filler. The pollution of runoff by turbidity and suspended solids influenced the term of filtration. Runoff, which has the lower contamination before filter (SS~10 mg/L,  $C_0$  ~50 mg/L), was filtered 80 minutes longer in rate of 30 m/h, than the more polluted runoff (SS~25 mg/L,  $C_0$  ~150 mg/L). In filtration of runoff (SS~25 mg/L,  $C_0$  ~150 mg/L) in the rate of 30 m/h, after 100 minutes, in the samples of filtrate, petroleum products' concentration of 5 mg/L was exceeded, meanwhile, the efficiency of absorption was up to 96.6 %. In decrease of filtration rate of runoff (SS~40 mg/L,  $C_0$  ~100 mg/L) from 30 to 15 m/h, the efficiency of absorption increased from 98.6 to 99.5 %, although the initial concentration of petroleum products increased from 100 to 158.27 mg/L. To filtrate more polluted storm water (such as SS~40 mg/L,  $C_0$  ~100 mg/L) is beside the purpose, because of the rapid filter and the obstruction of the grid. The derived sorption and hydraulic properties of the fibroil can be used to evaluate the efficiency of operative existing storm water treatment plants as well as to design new facilities.

#### REFERENCES

1. Pitt, M.; Brown, A.; Smith, A. 2002. Waste management at airports, *Facilities* 20(5/6): 198-207.
2. Deschamps, G.; Caruel, H.; Borredon, M.; Albasi, C.; Riba, J.; Bonnin, C.; Vignoles, C. 2003. Oil removal from water by sorption on hydrophobic cotton fibers. 2. Study of sorption properties in dynamic mode, *Environmental Science Technology* 37: 5034-5039.
3. Ke, L.; Yu, K.; Wong, Y.; Tam, N. 2005. Spatial and vertical distribution of polycyclic aromatic hydrocarbons in mangrove sediments, *Science of the Total Environment* 340: 177-187.
4. Garg, V.; Amit, M.; Gupta, R. 2004. Basic dye (methyl blue) removal from simulated wastewater by adsorption using Indian Rosewood sawdust a timber industry waste, *ScienceDirect. Dyes and Pigments* 63: 243-250.
5. Mažeikienė, A.; Rimeika, M.; Valentukevičienė, M.; Oškinis, V.; Paškauskaitė, N.; Brannvall, E. 2005. Removal of petroleum products from water using natural sorbent zeolite, *Journal of Environmental Engineering and Landscape Management* 13(4): 187-192.
6. Brannvall, E.; Mažeikienė, A.; Valentukevičienė, M. 2006. Experimental research on sorption of petroleum products from water by natural clinoptilolite and vermiculite, *Geologija* 56: 5-12.
7. Wei, Q.; Mather, R.; Fotheringham, A.; Yang, R. 2003. Evaluation of non woven polypropylene oil sorbents in marine oilspill recovery, *Marine Pollution Bulletin* 46 (6): 780-783.
8. Annunciado, T.; Sydenstricker, T.; Amico, S. 2005. Experimental investigation of various vegetable fibers as sorbent materials for oil spills, *Marine Pollution Bulletin* 50 (11): 1340-1346.
9. Husein, M.; Amer, A.; Sawsan, I. 2008. Oil spill sorption using carbonized pith bagasse: trial for practical application, *Journal of Environmental Science and Technology* 5 (2): 233-242.
10. Rajakovic, V.; Aleksic, G.; Radetic, M.; Rajakovic, L. 2006. Efficiency of oil removal from real wastewater with different sorbent materials, *Journal of Hazardous Materials* 143: 494-499.
11. Davies-Colley, R.; Smith, D. 2001. Turbidity, suspended sediment, and water clarity: a review. *Journal of the American water resources association* 37(5): 1085-1101.
12. Rahmah, A.; Abdullah, M. 2010. Evaluation of Malaysian *Ceiba pentandra* (L.) Gaertn. for oily water filtration using factorial design, *Desalination*. doi: 10.1016/j.desal.2010.08.001.